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. acany or chomical recimency					
		STUDY MODULE D	ESCRIPTION FORM		
Name of the module/subject Co		Code 1010702221010702657			
Field of	study nical Technolog	v	Profile of study (general academic, practical (brak)		
	path/specialty	у	Subject offered in:	1 / 2 Course (compulsory, elective)	
Elective		es and Nanomaterials	Polish	obligatory	
Cycle of	· · · · · · · · · · · · · · · · · · ·		Form of study (full-time,part-time)		
Second-cycle studies		full-time			
No. of h	ours			No. of credits	
Lectur	e: 30 Classes	s: - Laboratory: 45	Project/seminars:	15 6	
Status o	of the course in the study	program (Basic, major, other)	(university-wide, from another	field)	
		(brak)		(brak)	
Education	on areas and fields of sci	ence and art		ECTS distribution (number and %)	
technical sciences			6 100%		
Resp	onsible for subje	ect / lecturer:			
ema tel. (Fac	. dr hab. Elżbieta Frąc nil: elzbieta.frackowiak 616653632 ulty of Chemical Tech Berdychowo 4 60-965	@put.poznan.pl nology			
		s of knowledge, skills and	d social competencies:	1	
1	Knowledge	Student should be familiar with the backgrounds of electrochemistry. Student should be familiar with the backgrounds of material chemistry. Student should be familiar with the backgrounds of physical chemistry.			
2	Skills	Student should be able to communicate in English. Student should be able to self-education.			
3	Social competencies	Student should understand the need of self-education in terms of reading literature recommended by lecturer. Student should understand the importance of working separately and as a part of team.			
Assu	mptions and obi	ectives of the course:	importance of working separate	ery and as a part of team.	
The stu	udents should get acq	uainted with the novel materials of vanced energy sources.	f power sources, conversion of	chemical energy into electrical	
	Study outco	mes and reference to the	educational results for	a field of study	
Knov	/ledge:			-	
1. Stud	lent is able to find the	differences between various mate	••	• -	
2. Student is able to schedule appropriate materials for energy conversion and storage - [K_W06,K_W07]					
Skills					
Student knows the pathway for selecting appropriate material for energy storage process - [K_U11,K_U15] Student windows and the great selecting appropriate material for energy storage process - [K_U104,K_U105]					
2. Student understands the mechanism of energy accumulation in different materials - [K_U21,K_U22]					
Social competencies:					
	lent is able to self-edu		01		
2. Student understands the need of self-development - [K_K02]					

Assessment methods of study outcomes			
Written exam after lectures.			
Course description			

3. Student understands the importance of the team-working - [K_K04]

Faculty of Chemical Technology

Examples of generation and storage of energy. Main characteristics of power sources (capacity, power, energy, etc). Ragone plot. Application of different materials for conversion of chemical energy into electrical one. Electrode/electrolyte interface in the various power sources. Performance of electrochemical capacitor. Supercapacitors: materials, pseudocapacitance, solvation-desolvation phenomena. Pseudocapacitive materials: conducting polymers, transition metal oxides, carbon materials with heteroatoms (nitrogen, oxygen). Electrolyte as a source of pseudocapacitane effects. Symmetric, asymmetric and hybrid systems. Principle of lithium-ion cell. Solid electrolyte interface. Novel generation of lithium-ion batteries. Advanced materials for new power sources. Ionic liquids as a new green electrolyte. Flow-redox systems. Fuel cells: materials, performance, different types of fuel cells. Photovoltaic cells. Dye-sensitized solar cells. Application of novel energy sources.

Basic bibliography:

- 1. Nanomaterials Handbook ed. Y. Gogotsi, Taylor and Francis, Florida, 2006
- 2. B. E. Conway, Electrochemical Supercapacitors? scientific fundamentals and technological applications, Kluwer Academic/Plenum, New York 1999.
- 3. Carbons for Electrochemical Energy Storage and Conversion Systems, F. Beguin, E. Frackowiak eds., CRC Press, Boca Raton, FL, USA, 2010

Additional bibliography:

Result of average student's workload

Activity	Time (working hours)
1. Lecture	30
2. Consultation to lecture	10
3. Laboratory classes (practice)	45
4. Consultation to laboratory	25
5. Project	15
6. Consultation to project	23
7. Exam	2

Student's workload

Source of workload	hours	ECTS
Total workload	150	6
Contact hours	135	0
Practical activities	60	0